

Syllabus checklist

MODULE 5 Equilibrium and Acid Reactions CH12–12

What happens when chemical reactions do not go through to completion?

- ☐ conduct practical investigations to analyse the reversibility of chemical reactions, for example:...
- ☐ model static and dynamic equilibrium and analyse the differences between open and closed systems (ACSCH079, ACSCH091)
- ☐ analyse examples of non–equilibrium systems in terms of the effect of entropy and enthalpy, for example:...
- ☐ investigate the relationship between collision theory and reaction rate in order to analyse chemical equilibrium reactions (ACSCH070, ACSCH094)

What factors affect equilibrium and how?

- ☐ investigate the effects of temperature, concentration, volume and/or pressure on a system at equilibrium and explain how Le Chatelier's principle can be used to predict such effects, for example:...
- ☐ explain the overall observations about equilibrium in terms of the collision theory (ACSCH094)
- ☐ examine how activation energy and heat of reaction affect the position of equilibrium

How can the position of equilibrium be described and what does the equilibrium constant represent?

- ☐ deduce the equilibrium expression (in terms of K_{eq}) for homogeneous reactions occurring in solution (ACSCH079, ACSCH096)
- ☐ perform calculations to find the value of K_{eq} and concentrations of substances within an equilibrium system, and use these values to make predictions on the direction in which a reaction may proceed (ACSCH096)
- ☐ qualitatively analyse the effect of temperature on the value of K_{eq} (ACSCH093)
- ☐ conduct an investigation to determine K_{eq} of a chemical equilibrium system, for example:...
- ☐ explore the use of K_{eq} for different types of chemical reactions, including but not limited to:...

How does solubility relate to chemical equilibrium?

- ☐ describe and analyse the processes involved in the dissolution of ionic compounds in water
- ☐ investigate the use of solubility equilibria by Aboriginal and Torres Strait Islander Peoples when removing toxicity from foods, for example:...
- ☐ conduct an investigation to determine solubility rules, and predict and analyse the composition of substances when two ionic solutions are mixed, for example:...
- ☐ derive equilibrium expressions for saturated solutions in terms of K_{sp} and calculate the solubility of an ionic substance from its K_{sp} value
- ☐ predict the formation of a precipitate given the standard reference values for K_{sp}

MODULE 6 Acid/Base Reactions CH12–13

What is an acid and what is a base?

- ☐ investigate the correct IUPAC nomenclature and properties of common inorganic acids and bases (ACSCH067)
- ☐ conduct an investigation to demonstrate the preparation and use of indicators as illustrators of the characteristics and properties of acids and bases and their reversible reactions (ACSCH101)
- ☐ predict the products of acid reactions and write balanced equations to represent:...
- ☐ investigate applications of neutralisation reactions in everyday life and industrial processes
- ☐ conduct a practical investigation to measure the enthalpy of neutralisation (ACSCH093)
- ☐ explore the changes in definitions and models of an acid and a base over time to explain the limitations of each model, including but not limited to:...

What is the role of water in solutions of acids and bases?

- ☐ conduct a practical investigation to measure the pH of a range of acids and bases
- ☐ calculate pH, pOH, hydrogen ion concentration ($[H^+]$) and hydroxide ion concentration ($[OH^-]$) for a range of solutions (ACSCH102)
- ☐ conduct an investigation to demonstrate the use of pH to indicate the differences between the strength of acids and bases (ACSCH102)
- ☐ write ionic equations to represent the dissociation of acids and bases in water, conjugate acid/base pairs in solution and amphiprotic nature of some salts, for example:...
- ☐ construct models and/or animations to communicate the differences between strong, weak, concentrated and dilute acids and bases (ACSCH099)
- ☐ calculate the pH of the resultant solution when solutions of acids and/or bases are diluted or mixed

How are solutions of acids and bases analysed?

- ☐ conduct practical investigations to analyse the concentration of an unknown acid or base by titration
- ☐ investigate titration curves and conductivity graphs to analyse data to indicate characteristic reaction profiles, for example:...
- ☐ model neutralisation of strong and weak acids and bases using a variety of media
- ☐ calculate and apply the dissociation constant (K_a) and pK_a ($pK_a = -\log_{10}(K_a)$) to determine the difference between strong and weak acids (ACSCH098)
- ☐ explore acid/base analysis techniques that are applied:...
- ☐ conduct a chemical analysis of a common household substance for its acidity or basicity (ACSCH080) , for example:...
- ☐ conduct a practical investigation to prepare a buffer and demonstrate its properties (ACSCH080)
- ☐ describe the importance of buffers in natural systems (ACSCH098, ACSCH102)

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MODULE 7 Organic Chemistry

CH12–14

How do we systematically name organic chemical compounds?

- ☐ investigate the nomenclature of organic chemicals, up to C8, using IUPAC conventions, including simple methyl and ethyl branched chains, including: (ACSCH127)
- ☐ explore and distinguish the different types of structural isomers, including saturated and unsaturated hydrocarbons, including: (ACSCH035)

How can hydrocarbons be classified based on their structure and reactivity?

- ☐ construct models, identify the functional group, and write structural and molecular formulae for homologous series of organic chemical compounds, up to C8 (ACSCH035) ...
- ☐ conduct an investigation to compare the properties of organic chemical compounds within a homologous series, and explain these differences in terms of bonding (ACSCH035)
- ☐ analyse the shape of molecules formed between carbon atoms when a single, double or triple bond is formed between them
- ☐ explain the properties within and between the homologous series of alkanes with reference to the intermolecular and intramolecular bonding present
- ☐ describe the procedures required to safely handle and dispose of organic substances (ACSCH075)
- ☐ examine the environmental, economic and sociocultural implications of obtaining and using hydrocarbons from the Earth

What are the products of reactions of hydrocarbons and how do they react?

- ☐ investigate, write equations and construct models to represent the reactions of unsaturated hydrocarbons when added to a range of chemicals, including but not limited to:...
- ☐ investigate, write equations and construct models to represent the reactions of saturated hydrocarbons when substituted with halogens

How can alcohols be produced and what are their properties?

- ☐ investigate the structural formulae, properties and functional group including:...
- ☐ explain the properties within and between the homologous series of alcohols with reference to the intermolecular and intramolecular bonding present
- ☐ conduct a practical investigation to measure and reliably compare the enthalpy of combustion for a range of alcohols
- ☐ write equations, state conditions and predict products to represent the reactions of alcohols, including but not limited to (ACSCH128, ACSCH136):...
- ☐ investigate the production of alcohols, including:...
- ☐ investigate the products of the oxidation of primary and secondary alcohols
- ☐ compare and contrast fuels from organic sources to biofuels, including ethanol

What are the properties of organic acids and bases?

- ☐ investigate the structural formulae, properties and functional group including:...
- ☐ explain the properties within and between the homologous series of carboxylic acids amines and amides with reference to the intermolecular and intramolecular bonding present
- ☐ investigate the production, in a school laboratory, of simple esters
- ☐ investigate the differences between an organic acid and organic base
- ☐ investigate the structure and action of soaps and detergents

What are the properties and uses of polymers?

- ☐ model and compare the structure, properties and uses of addition polymers of ethylene and related monomers, for example:...
- ☐ model and compare the structure, properties and uses of condensation polymers, for example:...

MODULE 8 Applying Chemical Ideas

CH12–15

How are the ions present in the environment identified and measured?

- ☐ analyse the need for monitoring the environment
- ☐ conduct qualitative investigations – using flame tests, precipitation and complexation reactions as appropriate – to test for the presence in aqueous solution of the following ions:...
- ☐ conduct investigations and/or process data involving:...
- ☐ conduct investigations and/or process data to determine the concentration of coloured species and/or metal ions in aqueous solution, including but not limited to, the use of:...

How is information about the reactivity and structure of organic compounds obtained?

- ☐ conduct qualitative investigations to test for the presence in organic molecules of the following functional groups:...
- ☐ investigate the processes used to analyse the structure of simple organic compounds addressed in the course, including but not limited to:...

What are the implications for society of chemical synthesis and design?

- ☐ evaluate the factors that need to be considered when designing a chemical synthesis process, including but not limited to:...